A Comparison of the Dissolution Properties of Two Types of HPMC (Hypromellose) Capsules Filled with Basic APIs in Acid Media II Makoto Aso, Michikazu Horie Qualicaps Co, Ltd.



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Purpose

- > There are two types of hypromellose capsules, i.e., manufactured by 'cold gelling' or 'thermal gelling' dipping process, using mold pins kept at room temperature or pre-heated at 50~100°C, respectively. In addition, the 'cold gelling' requires a gelling agent such as carrageenan. We had previously carried out comparative studies on these two types of hypromellose capsules. The 'cold gelling' capsules with carrageenan and KCI showed excellent mechanical strength, i.e., less brittle, and fast disintegration with less variation.¹⁾ The fast disintegration was thought to be derived from the unique microstructure in the 'cold gelling' capsule, i.e., seaisland structure.2)
- ➤ On the other hand, the paper by Ku et al. 3) claimed that 'cold gelling' hypromellose capsules with carrageenan could retard dissolution in acidic medium when two basic compounds with two different pKa were encapsulated, although details on APIs or excipients were not disclosed. They speculated that retardation was due to an interaction between carrageenan and these basic compounds. However, we showed the 'cold gelling' capsule, compared to the thermal gelling capsule, had even shorter rupture times, for the three structurally different basic APIs.⁴⁾ In this study, we further investigated dissolution properties of these two types of hypromellose capsules with other basic APIs.

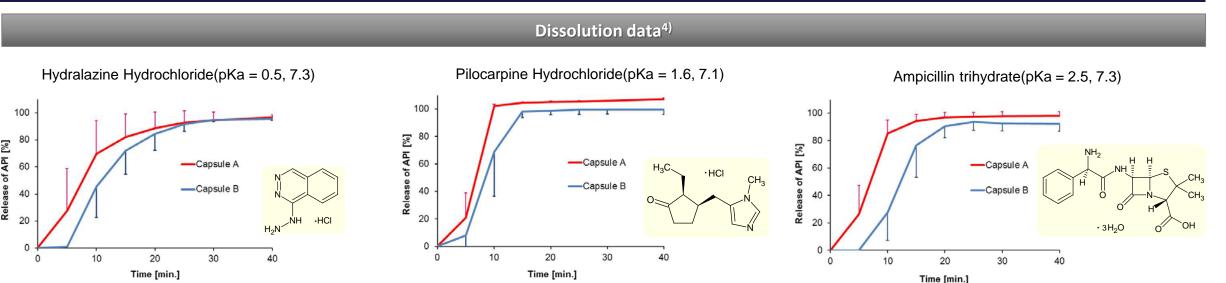
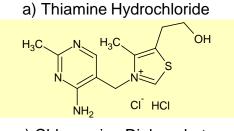


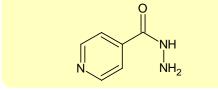
Figure 1. Dissolution profiles of basic APIs filled in HPMC capsules in 0.1N HCl at 37 °C.

Method

Table 1. Physicochemical and biopharmaceutical properties of basic APIs in dissolution tests.

АРІ	Form	BCS	рКа	Fill weight [mg]	Dissolution media	Solubility [mg/mL]	Sink ratio [-]	Paddle speed [rpm]	
Thiamine Hydrochloride	Salt	III ⁵⁾	4.85 ⁹⁾	50	0.1N HCI 34mM NaCI (JP 1st)	614	11,052	50	
Isoniazid	Base	I/III ⁶⁾	1.82, 3.52, 10.72 ⁹⁾	100		145	1,305		
Chloroquine Diphosphate	Salt	[7)	8.1, 10.07 ⁹⁾	100		347	3,123		
Chlorpheniramine Maleate	Salt	[8)	4.0, 9.2 ^{10,11)}	200		243	1,092		





b) Isoniazid

c) Chloroquine Diphosphate

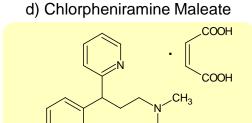


Figure 2. Chemical structure of basic APIs

submerged. Capsule contents: HPMC capsules were filled with each neat API.

Capsules: Size 1 hypromellose capsules were obtained from the Japanese market.

Equipment: Distek Dissolution Apparatus Model2100C; with a fiber-optic UV detection

Conditions: Paddle method at 50 rpm in acid medium of 900 mL 0.1N HCl, 34mM

system was employed. Data were collected from 6 samples and were

NaCl (1st fluid, JP17) at 37°C. JP sinkers were used to keep capsules

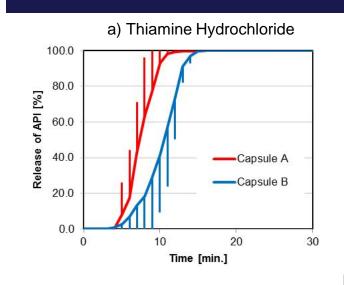
Capsule A: Quali-V®, 'cold gelling' with carrageenan as a gelling agent

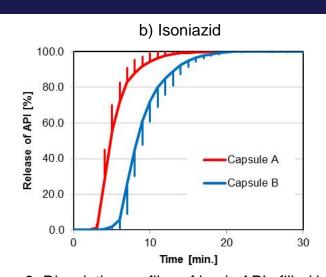
Capsule B: 'thermal gelling' without gelling agent

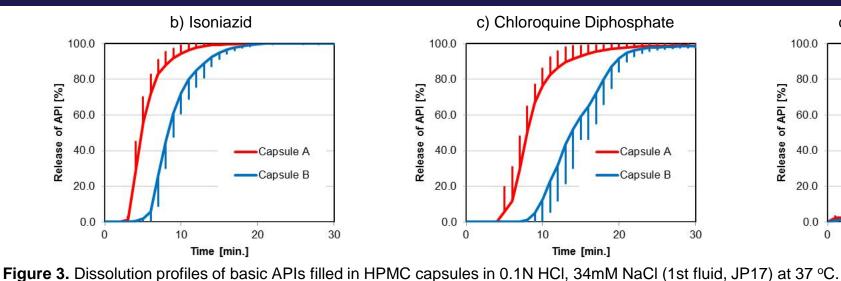
submitted for analysis.

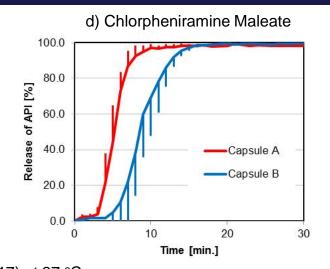
Dissolution tests

Results









- Dissolution tests showed that Capsule A compared to Capsule B had shorter rupture times and improved dissolution rates, for four structurally different APIs shown in **Figure 3**.
- In addition, the dissolution profiles of the Capsule A showed much less variation especially between 5 and 15 minutes.
- Results indicated no significant charge interactions between carrageenan and basic compounds in acid medium.

This study showed that dissolution rate from 'cold gelling' hypromellose capsules with carrageenan as a gelling agent, filled with basic APIs, showed no adverse effect in dissolution behavior in acid medium. Similar to dissolution behavior with acetaminophen in the previous studies, confirming that the excellent dissolution behavior of 'cold gelling' Capsule A with various types of basic APIs.

Conclusion

- 1) T. Uyama et al., Poster W4188, AAPS Annual Meeting, New Orleans, November 2010.
- 2) T. Usui et al., Poster M1326, AAPS Annual Meeting, Orlando, November 2015.
- 3) M. S. Ku et al., Int. J. Pharm., 416, 16, 2011.
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Reference

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- 8) C. Wu et al., Pharm. Res., 22, 11, 2005.
- 9) D. T. Manallack, SAR QSAR Environ. Res., 20, 611, 2009.
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